

Directional solidification and interface structure of BaTiO₃-CoFe₂O₄ eutectic

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The directional solidification and the interface structure of BaTiO₃-CoFe₂O₄ eutectic grown by the floating zone melting method were investigated. The microstructure of the eutectic were found to consist grains of lamellar or fibrous morphology. Two types of crystal structures of BaTiO₃ were observed for different growth conditions. The orientation relationships between BaTiO₃ and CoFe₂O₄ (the spinel-type crystal structure) are a) hcp BaTiO₃: (111) CoFe₂O₄ // (001) BaTiO₃ and (110) CoFe₂O₄ // (110) BaTiO₃, and b) tetra/cubic BaTiO₃: (001) CoFe₂O₄ // (001) BaTiO₃ and (100) CoFe₂O₄ // (100) BaTiO₃. The interface has misfit dislocations due to the accommodation of lattice mismatch. © 2000 Kluwer Academic Publishers

1. Introduction

The directional solidification of eutectics is a method used to produce an anisotropic composite directly from the melt [1, 2]. Production of composite materials having designed physical properties has been attempted with this method [3, 4].

The BaTiO₃-CoFe₂O₄ eutectic system is an interesting system because BaTiO₃ and CoFe₂O₄ are materials with large electric striction and magneto striction, respectively. Van den Boomgaard *et al.* have investigated the directional solidification of this system and shown that it is possibly applicable for new devices [5]. Application of this system will be achieved by the combination of magnetic and electric properties (magneto electric effect) through strain, a common property of the two oxides. The interface between the two oxides is important for the production of the magneto electric effect, because strain propagates through the interface and the bonding at the interface affects the efficiency of this effect. Though the directional solidification of the BaTiO₃-CoFe₂O₄ eutectic system has been studied, little information on the microstructure, including the interface, has been reported. The aim of this study was to investigate the growth morphology, orientation relationship, and interface structure of directionally solidified BaTiO₃-CoFe₂O₄ eutectics.

2. Experimental procedure

The eutectic composition was determined from the phase diagram of the BaTiO₃-CoFe₂O₄ system proposed by Van den Boomgaard, the given eutectic com-

position being 38 mol% CoFe₂O₄ for the BaTiO₃-CoFe₂O₄ eutectic [5]. Two kinds of blended raw powders, namely, (1) BaTiO₃ and CoFe₂O₄, and (2) Fe₂O₃, TiO₂, CoO (99.99% purity) and BaO (99.8% purity), were used in the present experiment. Powders were weighed, blended, and pressed into a rod with a diameter of 5 mm and a length of 80 mm. This rod was sintered at 1470 K for 3.6 ks. Directional solidification was conducted by the floating zone melting method using a single ellipsoid image furnace [6, 7]. Starting with the powders of (1), samples were grown in air at a growth rate of 10 to 90 mm/h. When powders of (2) were used as starting materials, directional solidification was performed by the same procedure as mentioned above, except that an atmosphere of 2.5% O₂ in N₂ was employed during solidification.

Directionally solidified eutectics were cut and polished perpendicular to the direction of growth. Qualitative chemical analysis of the phases was performed using EPMA (Electron Probe Micro Analyzer) on a polished surface of directionally solidified eutectics. Thin foil was prepared by sectioning the directionally solidified eutectics. The sections were ion-beam thinned in argon at 5 kV at an angle of 30 degrees. The resulting foil was examined by H-800 and JEOL-1250 transmission electron microscopy.

3. Results and discussion

Fig. 1 shows optical photographs of the directionally solidified BaTiO₃-CoFe₂O₄ eutectic grown in air

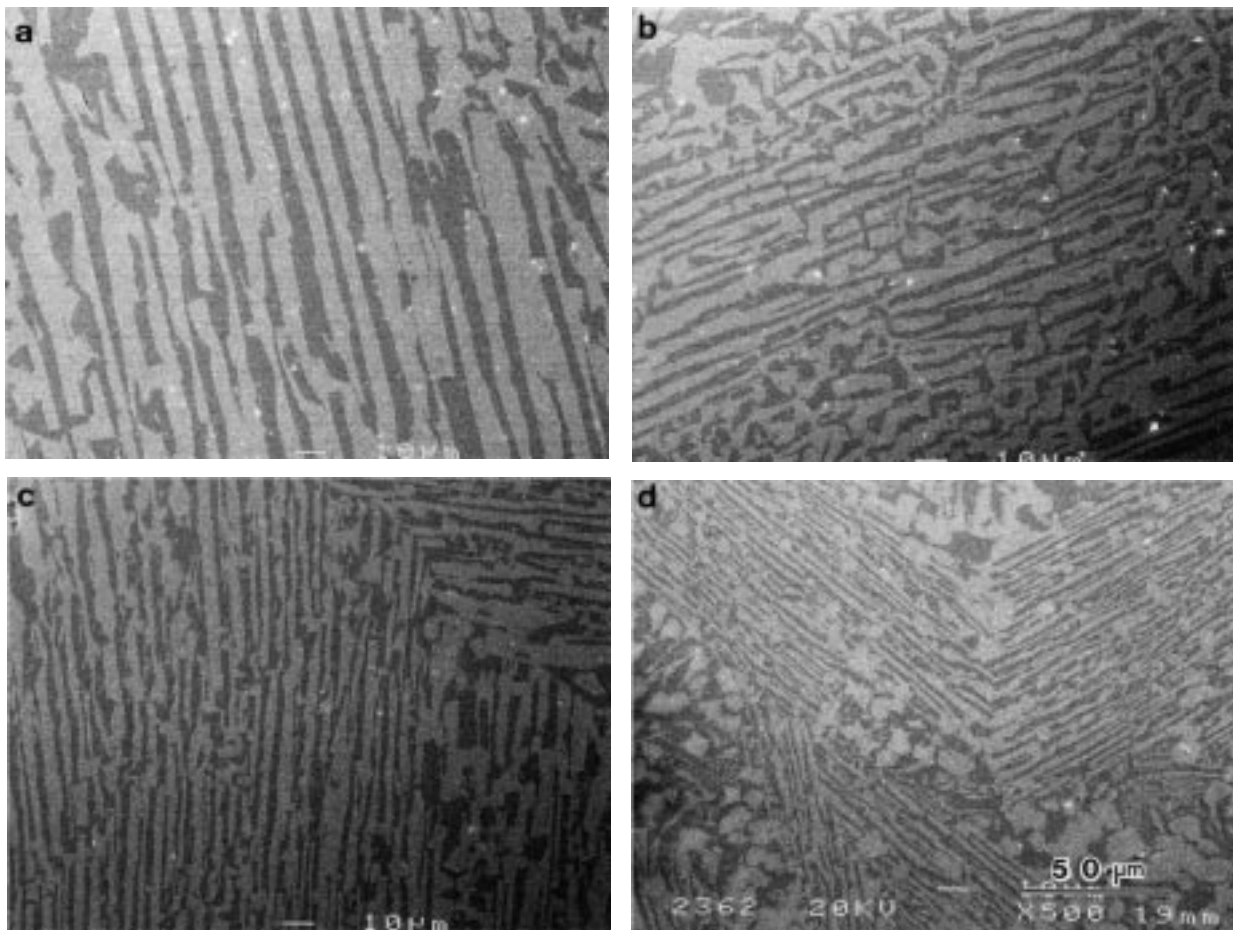


Figure 1 Optical photographs of samples of the directionally solidified BaTiO₃-CoFe₂O₄ eutectic grown in air. The growth rate: 10 mm/h (a), 20 mm/h (b), 40 mm/h (c) and 90 mm/h (d).

using the starting materials (1) of mixed BaTiO₃ and CoFe₂O₄ powders. The surface observed was that perpendicular to the direction of growth. The growth morphology of this system is a plate structure, i.e., a structure with lamellae. The structure of directional solidification was retained up to a growth rate of 90 mm/h without any cell structure. The relationship between the spacing of lamellae and the growth rate is shown in Fig. 2. The smaller the spacing of the lamellae is, the larger the growth rate is. In directionally solidified eutectics, the spacing between the two phases is plotted with the inverse of the square root of the growth rate, revealing the existence of a linear relationship between the two phases, which means that the relation $\lambda^2 R = \text{constant}$ holds in this system as shown in Fig. 2, where λ and R indicate the spacing between the two phases and the growth rate, respectively.

Chemical analysis of both phases in the BaTiO₃-CoFe₂O₄ eutectic was carried out by employing an EPMA as shown in Fig. 3. The bright area in Fig. 1 was identified as the BaTiO₃ phase consisting almost entirely of BaTiO₃, though a small amount of Fe was detected. On the other hand, the dark area in Fig. 1 was coincident with the CoFe₂O₄ phase and a small amount of Ti was detected in it.

Fig. 4 is an electron diffraction pattern taken from the interface of a directionally solidified eutectic grown at a rate of 20 mm/h in air. The two phases of the eutectic were confirmed as being BaTiO₃ with a hcp struc-

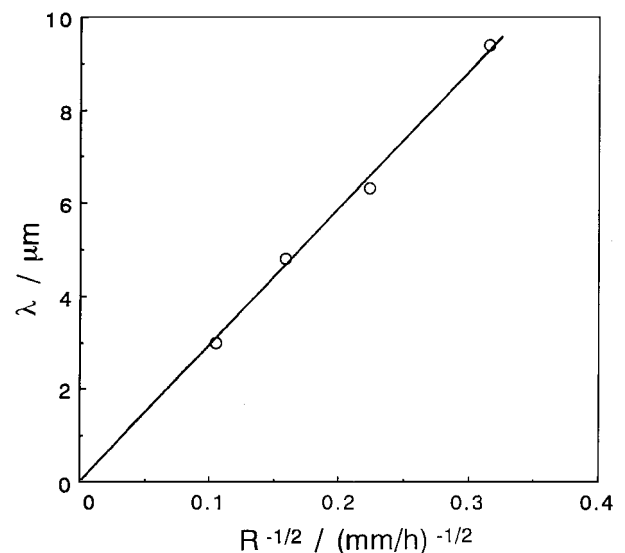


Figure 2 Relationship of spacing between the two phases shown by the inverse of the square root of the growth rate.

ture and CoFe₂O₄ with the spinel-type structure, respectively. If we use CoFe₂O₄ as a standard material for estimation of lattice parameters, the lattice parameters of the hcp BaTiO₃ can be determined from the electron diffraction pattern shown in Fig. 4. From this procedure, the lattice parameters of the hcp BaTiO₃ were $a = 0.575$ nm and $c = 1.406$ nm. These values, which are very close to those given in the references

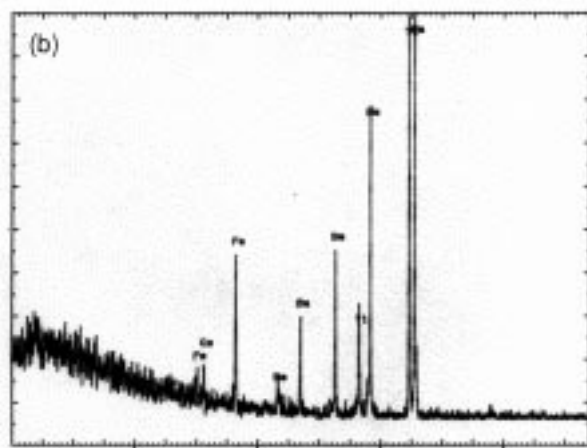
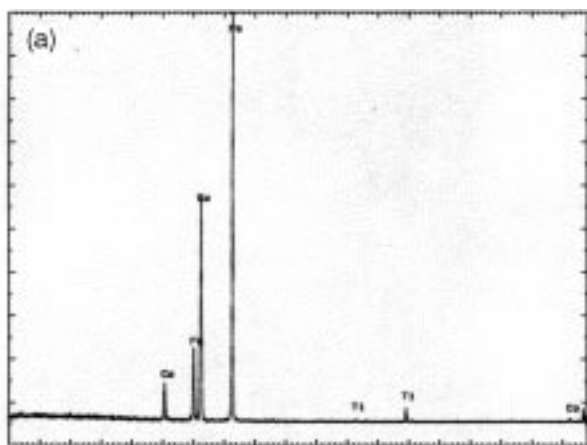


Figure 3 Chemical analysis of both phases in the BaTiO₃-CoFe₂O₄ eutectic by employing an EPMA. The dark and bright area in Fig. 1 was given by (a) and (b), respectively.

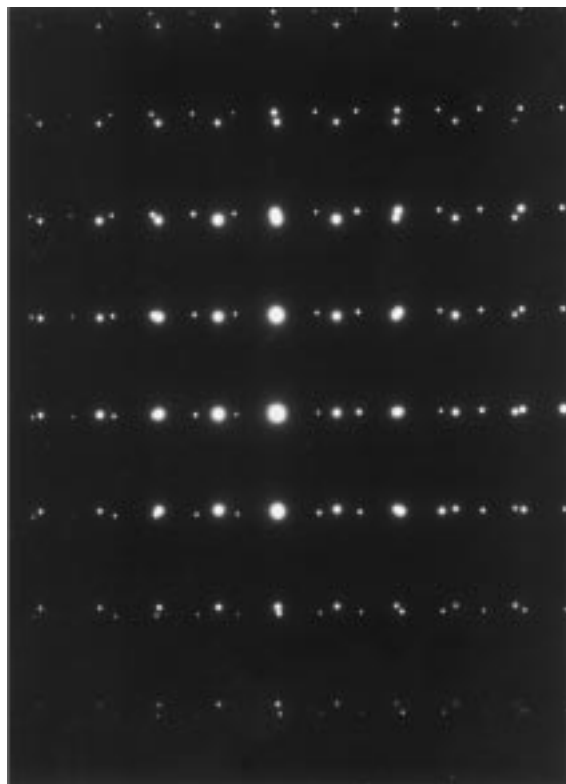
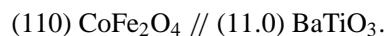
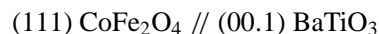


Figure 4 Electron diffraction pattern taken from the interface of a directionally solidified eutectic grown at a rate of 20 mm/h in air.

[8–10], suggest that the hcp BaTiO₃ obtained by the present experiment has the same structure as that previously reported. It should be noted that the hcp phase of BaTiO₃ is stabilized when Fe is a solute in BaTiO₃. In the present experiment, Fe was detected by EPMA analysis as shown in Fig. 3.

The orientation relationship between spinel-type CoFe₂O₄ and hcp BaTiO₃ can be represented as follows:

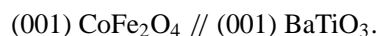
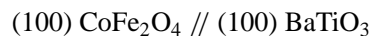


The interface plane, which has a lamellar structure, was formed by (111) CoFe₂O₄ // (00.1) BaTiO₃.

The high resolution lattice image of the interface between CoFe₂O₄ and BaTiO₃ is shown in Fig. 5, in which the orientation relationship is given as above, and the observed direction, i.e., the direction of the beam, is [112]CoFe₂O₄ // [10 $\bar{1}$ 0]hcp BaTiO₃. The interface was relatively flat and formed by (111) CoFe₂O₄ // (00.1) BaTiO₃ microscopically. The continuity of the lattice planes across the interface, in this case the (110) of CoFe₂O₄ and (11.0) hcp BaTiO₃, was extremely good, the misfit between them being only 3.6%. The misfit of 3.6% introduced misfit dislocations every 28 layers of the BaTiO₃, because the spacing of the lattice plane of CoFe₂O₄ is a little smaller than that of the hcp BaTiO₃.

Fig. 6 shows an optical photograph of a sample perpendicular to the growth direction of the directionally solidified BaTiO₃-CoFe₂O₄ eutectic. The starting materials of (b), in which the concentration of TiO₂ was a little greater than that of (a), were used to obtain tetragonal BaTiO₃. The growth condition was a growth rate of 20 mm/h in 2.5% O₂ in N₂. The growth morphology after directional solidification was a fibrous structure with dendrites.

Fig. 7 is an electron diffraction pattern from the interface of the directionally solidified eutectic grown at a rate of 20 mm/h in an atmosphere of 2.5% O₂-N₂. The growth direction in both the CoFe₂O₄ and BaTiO₃ phases is [001]. It is difficult to confirm a phase as being BaTiO₃ with a tetragonal or cubic structure, because the tetragonality of BaTiO₃ is too small to detect from the electron diffraction pattern ($c/a = 1.01$). Nevertheless, observation of the cubic BaTiO₃ suggests that a tetragonal BaTiO₃ phase may exist below 380 K. The orientation relationship between spinel CoFe₂O₄ and cubic/tetragonal BaTiO₃ was



The interface planes consisting of a fiber, were almost all (100) CoFe₂O₄ planes, which were parallel to (100) BaTiO₃ planes. These results suggest that the crystal structure of BaTiO₃ is determined by three factors, i.e., oxygen partial pressure during solidification, the addition of a small amount of TiO₂, and the growth direction of the eutectics.

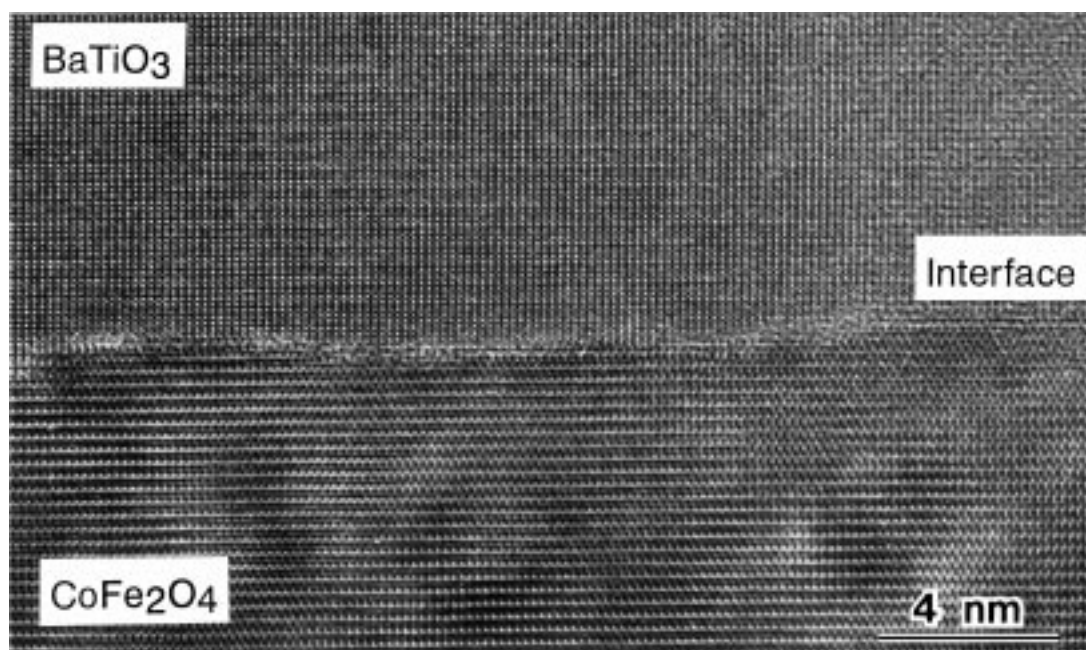


Figure 5 High resolution lattice image of the interface between CoFe_2O_4 and BaTiO_3 . Sample was the same as that in Fig. 4.



Figure 6 Optical photograph of the directionally solidified BaTiO_3 - CoFe_2O_4 eutectic grown under conditions of a growth rate of 20 mm/h in 2.5% O_2 in N_2 .

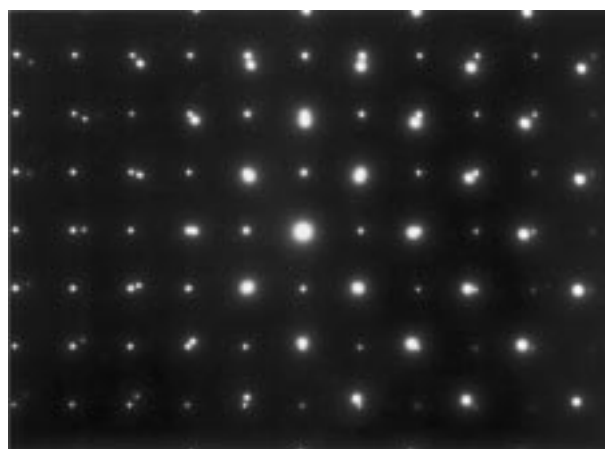


Figure 7 Electron diffraction pattern of the interface in the directionally solidified eutectic grown at a rate of 20 mm/h in a 2.5% O_2 - N_2 atmosphere.

The high resolution lattice image of the interface between CoFe_2O_4 and cubic/tetra BaTiO_3 is shown in Fig. 8, in which the growth direction, i.e., the beam direction, was $[001]\text{CoFe}_2\text{O}_4 // [001]$ cubic/tetra BaTiO_3 . The interface was relatively flat and formed by $(100)\text{CoFe}_2\text{O}_4 // (100)\text{BaTiO}_3$ microscopically. The continuity of lattice planes across the interface, in this case of (010) of CoFe_2O_4 and (010) cubic/tetra BaTiO_3 , was good, the misfit between them being only 4.9%. It should be noted that the misfit of the spinel-type $\text{CoFe}_2\text{O}_4/\text{hcp BaTiO}_3$ interface is smaller than that of the spinel-type $\text{CoFe}_2\text{O}_4/\text{cubic BaTiO}_3$ interface. The misfit of 4.9% introduces misfit dislocations every 20 layers of the BaTiO_3 . The contrast at the interface shown in Fig. 8 seems to suggest that the some kinds of solutes were segregated at the interface.

The electrical resistance measured along the direction of growth of the cubic/tetra BaTiO_3 - CoFe_2O_4 eutectic was high at 5×10^8 ohm, i.e., it was an insulator. It was almost constant with the increase of temperature. Fig. 9 shows the temperature dependence of the electric resistance of the hcp BaTiO_3 - CoFe_2O_4 eutectic measured along the direction of growth. As can be seen from this figure, the resistance decreased with the increase of temperature, i.e., the eutectic is a semiconductor. Jonker has investigated the electrical resistivities of $\text{Co}_{3-x}\text{Fe}_x\text{O}_4$ near the composition of CoFe_2O_4 [11]. When the composition, x , is less than 1.99 (in Co rich samples), the resistivities are on the order of 10^6 - 10^8 ohm cm. In the present case, the CoFe_2O_4 phase becomes a Co-rich phase, resulting from the dissolution of Fe in the BaTiO_3 phase, and then shows a high resistance of 10^8 ohm. Thus it may be concluded that the semiconducting behavior of the hcp BaTiO_3 - CoFe_2O_4 eutectic is due to the semiconductor hcp BaTiO_3 .

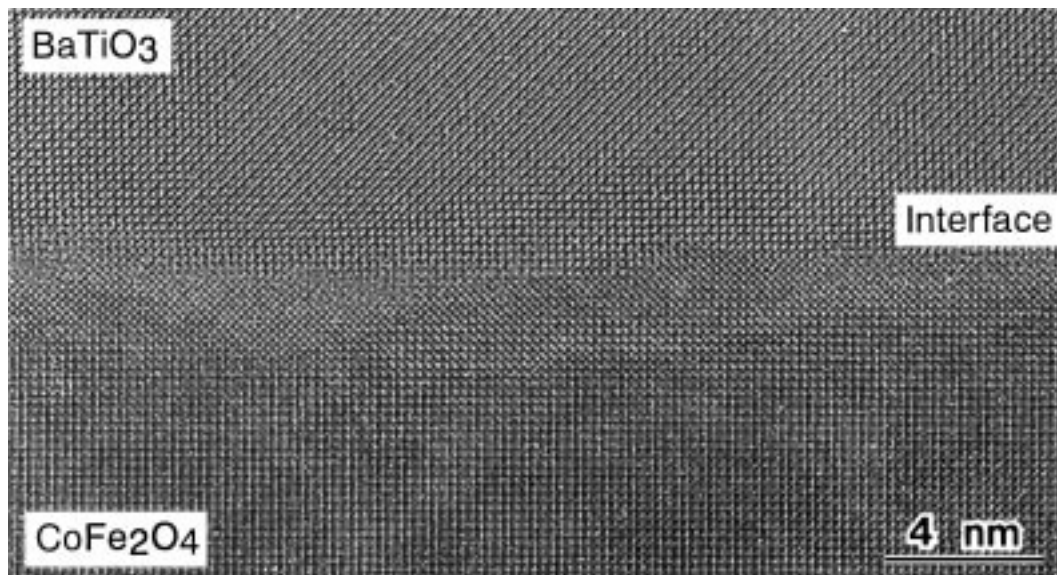


Figure 8 High resolution lattice image of the interface between CoFe_2O_4 and cubic/tetra BaTiO_3 .

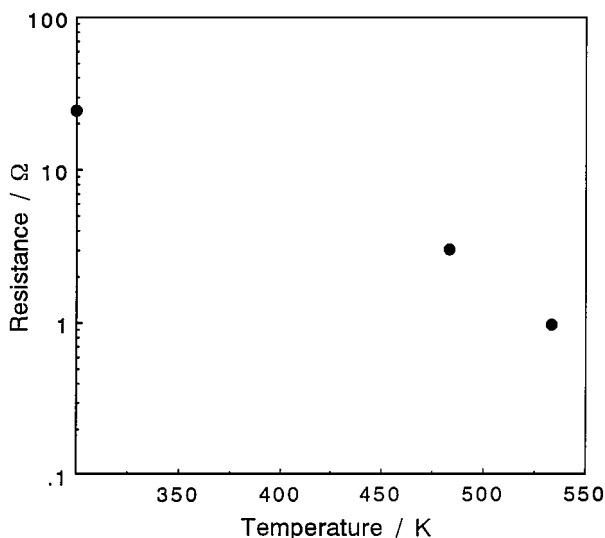


Figure 9 Temperature dependence of the electric resistance of the hcp BaTiO_3 - CoFe_2O_4 eutectic measured along the direction of growth.

4. Conclusions

The directional solidification and the interface structure of BaTiO_3 - CoFe_2O_4 eutectic were investigated by the floating zone melting method and following results were obtained.

(1) The micro structure of the eutectic consisted of grains of lamellar or fibrous morphology.

(2) The orientation relationships between BaTiO_3 and CoFe_2O_4 (the spinel type crystal structure) are a) hcp

BaTiO_3 : (111) CoFe_2O_4 // (00.1) BaTiO_3 and (110) CoFe_2O_4 // (11.0) BaTiO_3 , b) tetra/cubic BaTiO_3 : (001) CoFe_2O_4 // (001) BaTiO_3 and (100) CoFe_2O_4 // (100) BaTiO_3 .

(3) The interfaces had misfit dislocations due to the accommodation of lattice mismatch.

(4) The semiconducting behavior of the hcp BaTiO_3 - CoFe_2O_4 eutectic may be caused by the existence of the hcp BaTiO_3 due to the dissolution of Fe.

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